

## Section 5.6: cont'd

Monday, February 16, 2015

12:31 PM

According to Monday Magazine, "one rat can create 15,000 descendants in a single year." Assuming this is true

- calculate the doubling time for Victoria's rat population
- how long would it take for that one rat to have one million descendants instead?

$$a) \quad A = A_0 e^{rt}$$

$$2A_0 = A_0 e^{rt}$$

after one doubling time  
↳ now divide both sides by  $A_0$

$$2 = e^{rt}$$

↳ problem! two variables!

but how do we find  $r$ ?

use that 1 rat  $\rightarrow$  15000 in one year

$$A = A_0 e^{rt}$$

$$15000 = 1 e^{r \cdot 1}$$

$$15000 = e^r$$

$$\ln 15000 = \ln e^r$$

$$\ln 15000 = r \quad \text{or} \quad r \approx 9.615805^*$$

\* don't round too early - keep at least 4-5 sig figs

now return to doubling time

$$A = A_0 e^{rt}$$

$$2A_0 = A_0 e^{rt}$$

$$2 = e^{rt}$$

$$2 = e^{9.615805 t}$$

$$\ln 2 = 9.615805 t$$

$$t = \frac{\ln 2}{9.615805}$$

$$= 0.072084$$

$$= 0.072 \text{ years}$$

or convert to days

$$t = 26.3 \text{ days}$$

$$\times \left( \frac{365 \text{ days}}{1 \text{ year}} \right)$$

b)

$$A = A_0 e^{rt}$$

$$10^6 = 1 e^{9.615805 t}$$

$$\ln 10^6 = 9.615805 t$$

$$t = \frac{\ln 10^6}{9.615805} \approx 1.4 \text{ years}$$

## exponential decay:

Scientists at the Kellogg Radiation Lab are studying the radioactive isotope Cesium-137. The **half-life** of Cs-137 was found to be 57 days. If initially the lab had 1.5 g of Cs-137, how much was left after exactly 3 weeks had elapsed?

finding  $r$ :

$A_0 \rightarrow \frac{1}{2}A_0$  after 57 days

$$A = A_0 e^{-rt}$$

$$\frac{1}{2}A_0 = A_0 e^{-r \cdot 57}$$

$$\frac{1}{2} = e^{-57r}$$

$$\ln \frac{1}{2} = \ln e^{-57r}$$

$$\ln \frac{1}{2} = -57r$$

$$r = \frac{\ln \frac{1}{2}}{-57} \approx 0.0121605$$

now find  $A$ :

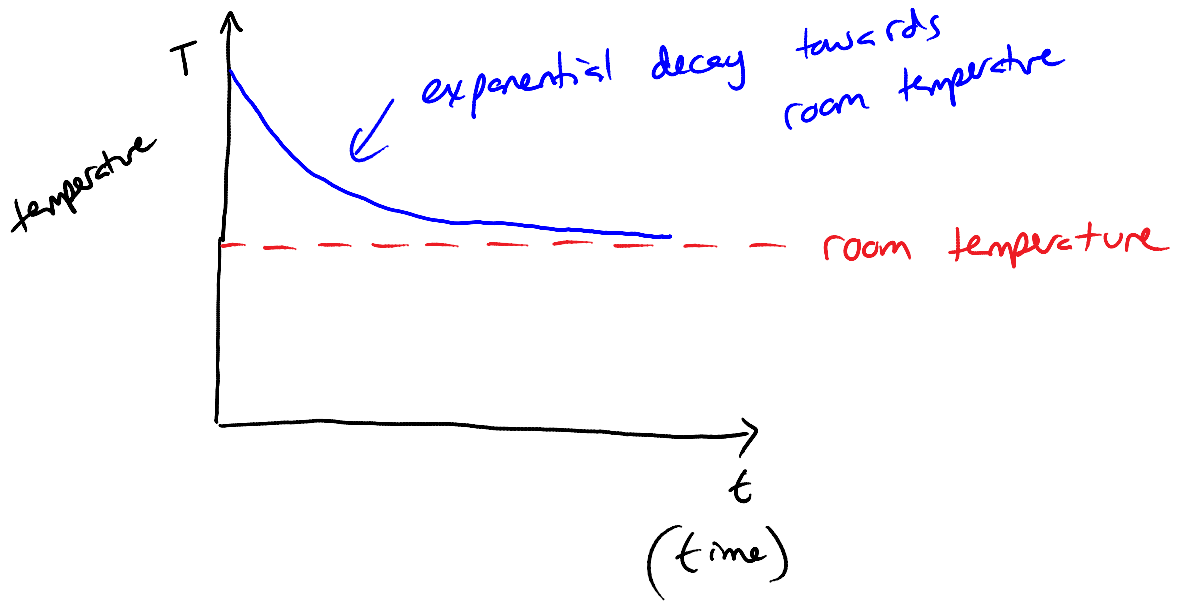
$$A = A_0 e^{-rt}$$

$$= 1.5 e^{-0.0121605(21)}$$

$$= 1.161944$$

$$= 1.2 \text{ g}$$

## Newton's Law of Cooling:



$$T = T_0 + (T_i - T_0) e^{-kt}$$

↗  
 $T(t)$   
 function  
 notation:  
 Temp is a  
 function of  
 time

$T$  = temp of object at time  $t$   
 $T_0$  = room temperature  
 $T_i$  = initial temp of object  
 $k$  = rate  
 $t$  = time elapsed

A batch of doughnuts is deep-fried at  $175^\circ\text{C}$  and then taken out and allowed to cool on the counter. When the room temperature is  $25^\circ\text{C}$ , the doughnuts take 5 minutes to cool to  $100^\circ\text{C}$ . If Homer can only wait 10 minutes before eating a doughnut and he'll burn his mouth if the

doughnut is hotter than  $50^{\circ}\text{C}$ , will he be able to munch doughnuts happily ever after?

$$T = T_0 + (T_1 - T_0) e^{-kt}$$

$T_0$  = room temp

$T_1$  = initial temp of object

find  $k$ :  $T_1 = 175^{\circ}\text{C}$   $\rightarrow$   $T = 100^{\circ}\text{C}$  after 5 min  
 $T_0 = 25^{\circ}\text{C}$

$$100 = 25 + (175 - 25) e^{-k \cdot 5}$$

$$75 = 150 e^{-5k}$$

$$\frac{1}{2} = e^{-5k}$$

$$\ln \frac{1}{2} = -5k$$

$$k = -\frac{\ln \frac{1}{2}}{5} \approx 0.138629$$

find  $T$  at 10 min: (could also find  $t$  for which  $T = 50^{\circ}\text{C}$ )

$$\begin{aligned} T &= T_0 + (T_1 - T_0) e^{-kt} \\ &= 25 + (175 - 25) e^{-0.138629(10)} \\ &= 62.5^{\circ}\text{C} \end{aligned}$$

D'oh! Homer will burn his math.

note:

$$75 = 150 e^{-sk}$$

note: better to  
divide by  
150 first

$$\begin{aligned}\ln 75 &= \ln 150 e^{-sk} \\ &= \ln 150 + \ln e^{-sk}\end{aligned}$$

$$\ln 75 - \ln 150 = -sk \ln e$$

$$\ln 75 - \ln 150 = -sk$$



$$\ln \frac{75}{150}$$

moral of story:  
divide by coeff first